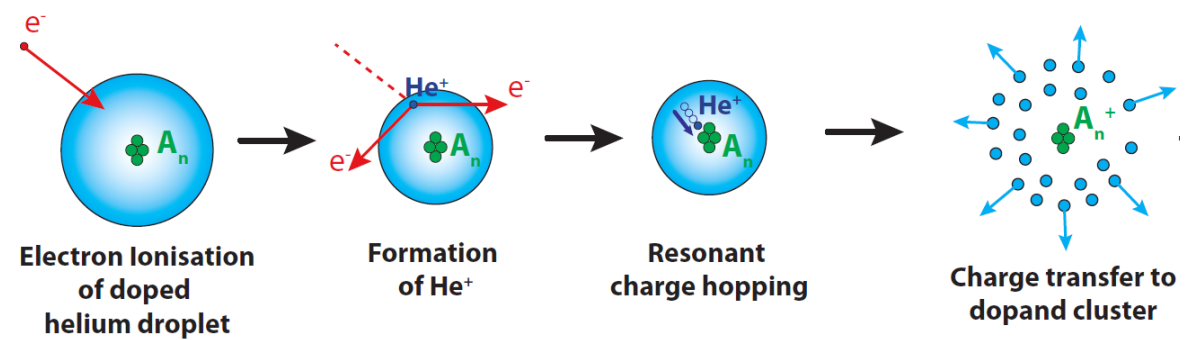
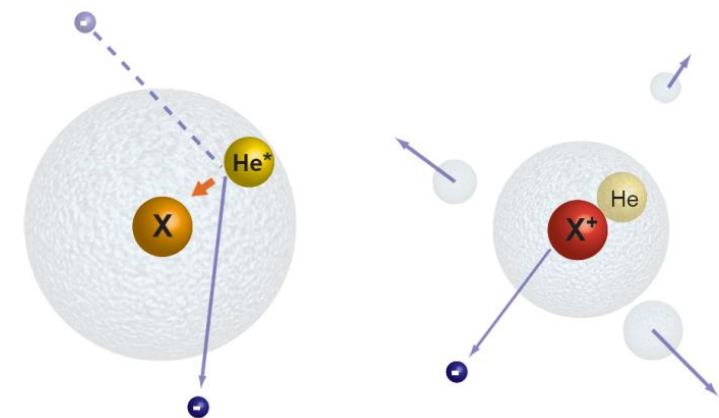


Detection of Negative Charge Carriers in
Superfluid Helium Droplets:
The Metastable Anions He^{*-} and He_2^{*-}

J. Postler, University of Innsbruck

The Environment: Helium Nano Droplets

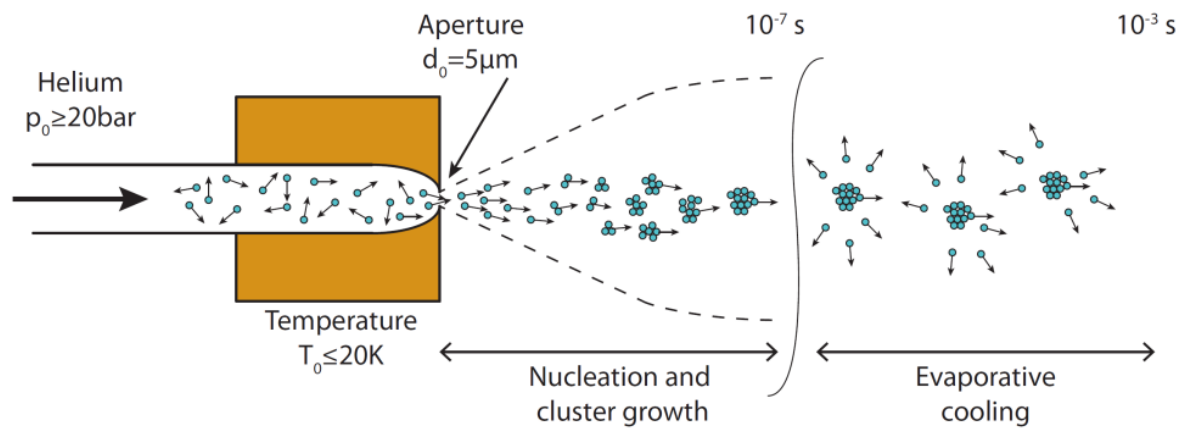
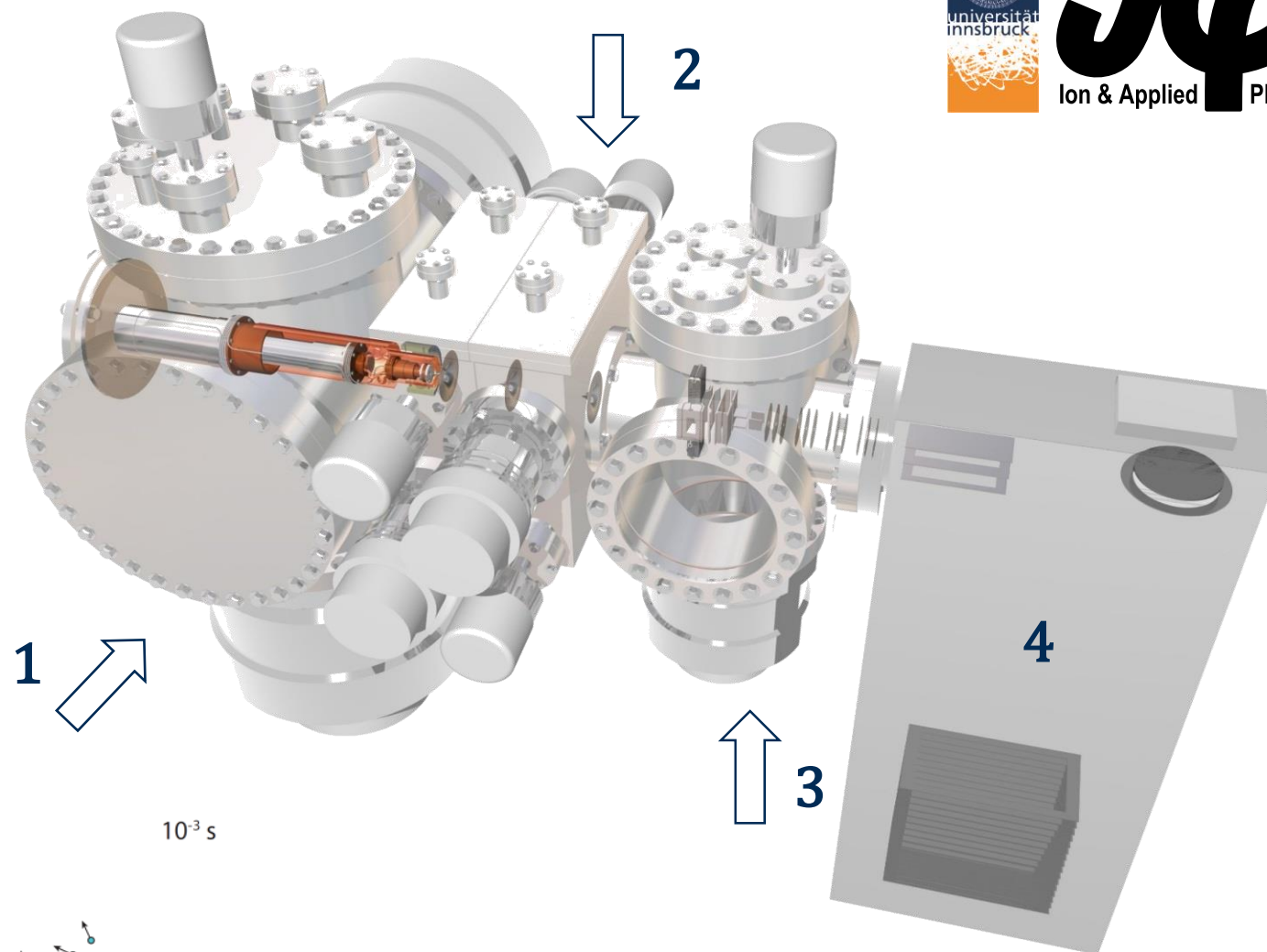
- HND provide a unique environment:
 - 0.37 K weakly interacting matrix
 - Cooling by evaporation of helium
 - Superfluidity
 - Transparent in wide range of wavelengths (IR to UV)
- Used for low temperature chemistry
- Several ionization mechanisms for dopants
 - $X + \text{He}^+ \rightarrow X^+ + \text{He}$
 - $X + \text{He}^* \rightarrow X^+ + \text{He} + e^-$
 - $X + e^- \rightarrow X^-$



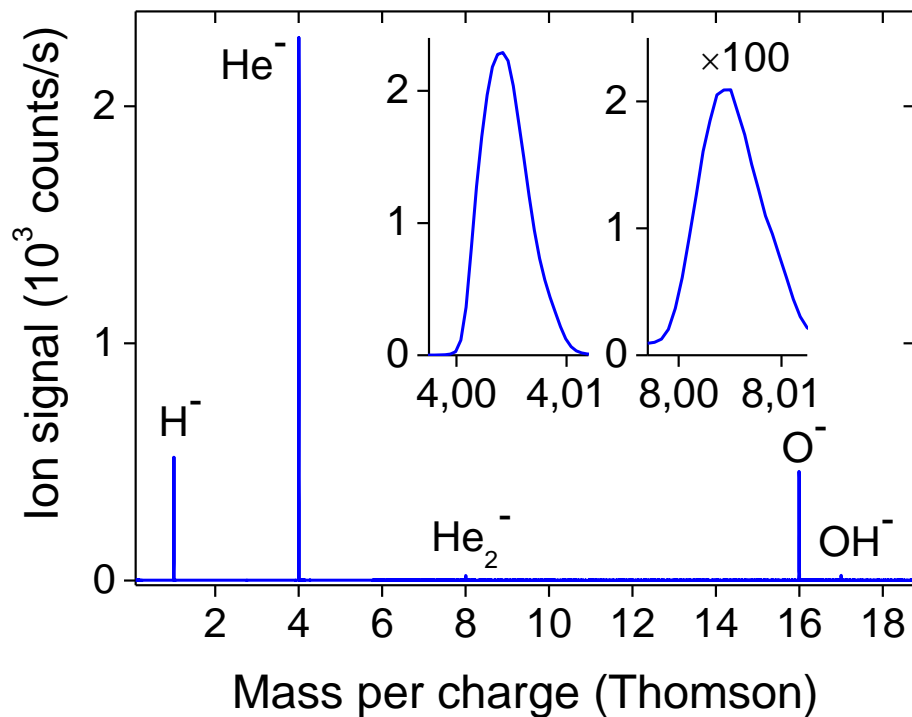
Apparatus

Cluster-Source Time-Of-Flight – CLUSTOF

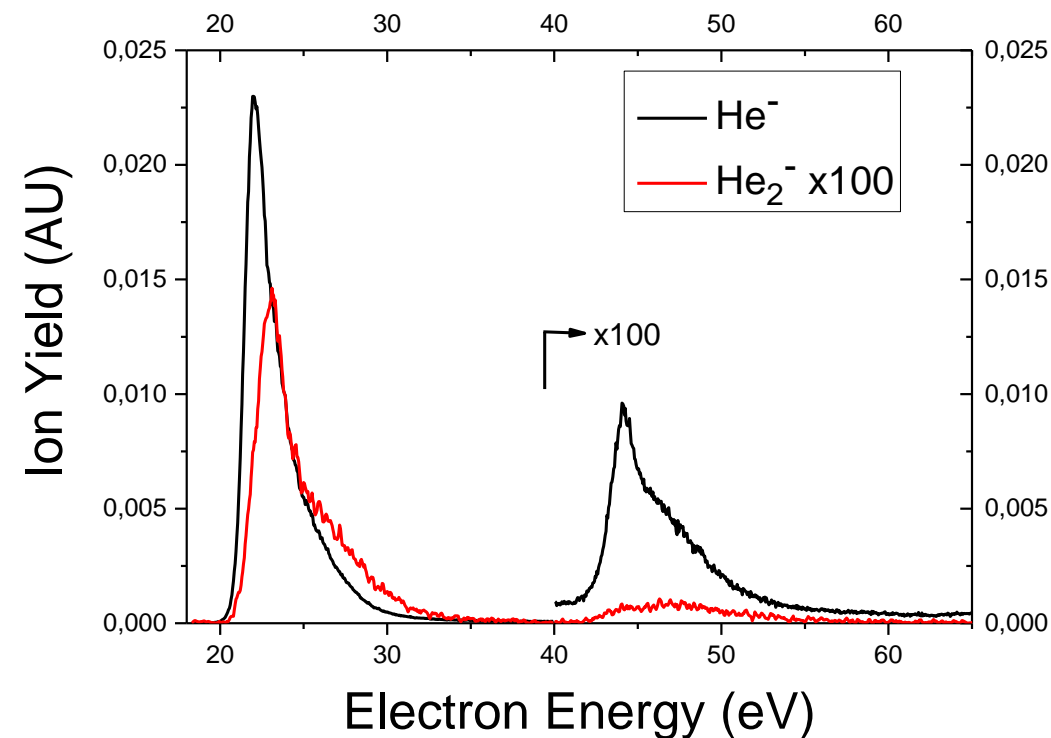
- 1) He-Cluster source
- 2) Pickup-Chambers
- 3) Ion source
- 4) TOF



He^{*-} and He_2^{*-}



- Clearly visible in the mass spectra
- Dimer about 100 times smaller
- (measured at 22 eV)



- Distinct peaks around 22 to 25 eV
- Repetition due to inelastic scattering of e⁻ at roughly 20 eV more energy

He^{*-} and He₂^{*-}

	He ^{*-}	He ₂ ^{*-}
First metastable state	1s2s2p ⁴ P	1σ _g ² 1σ _u 2σ _g 2π _u ⁴ Π _g
Related state	1s2s ³ S	1σ _g ² 1σ _u 2σ _g a ³ Σ _u ⁺
Binding energy [5, 6]	77 meV	0.23 eV
Lifetime [1, 2]	359 ± 0.9 μs (j = 5/2) 10 μs (j = 3/2; 1/2)	135 ± 15 μs
Year of Detection [3, 4]	1939	1984

[1] H. T. Schmidt et al., in International Conference on Photonic, Electronic and Atomic Collisions (IOP Publishing, 2011).

[2] T. Andersen et al., J. Phys. B **27**, 1135 (1994).

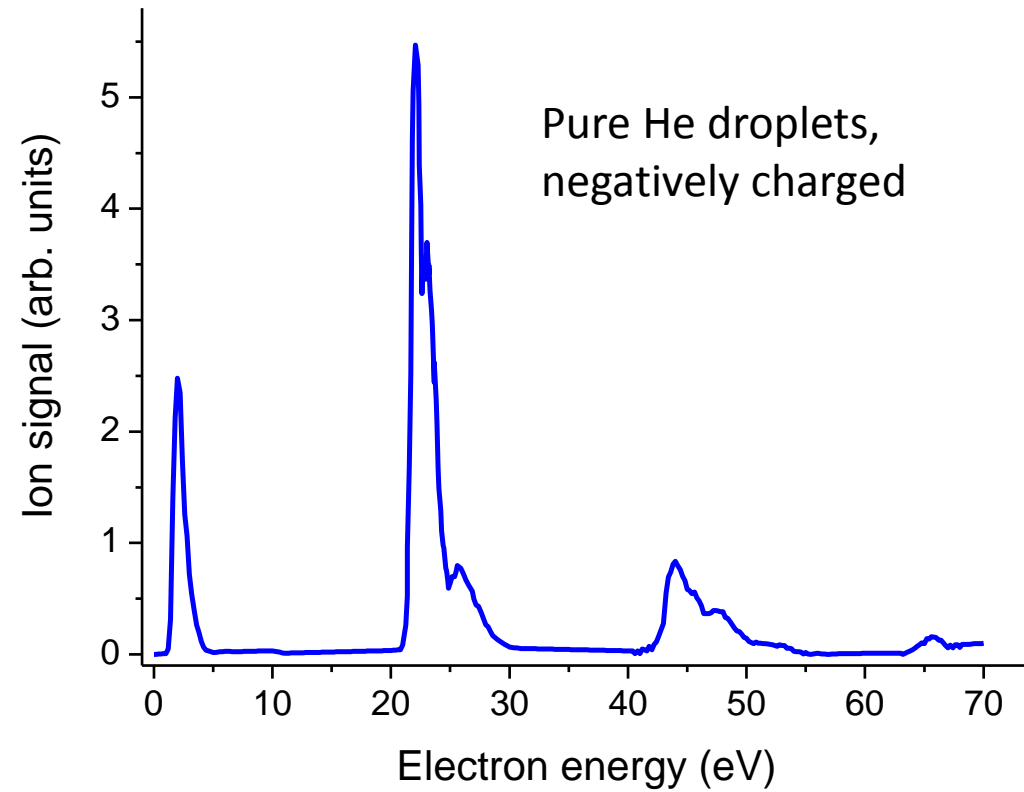
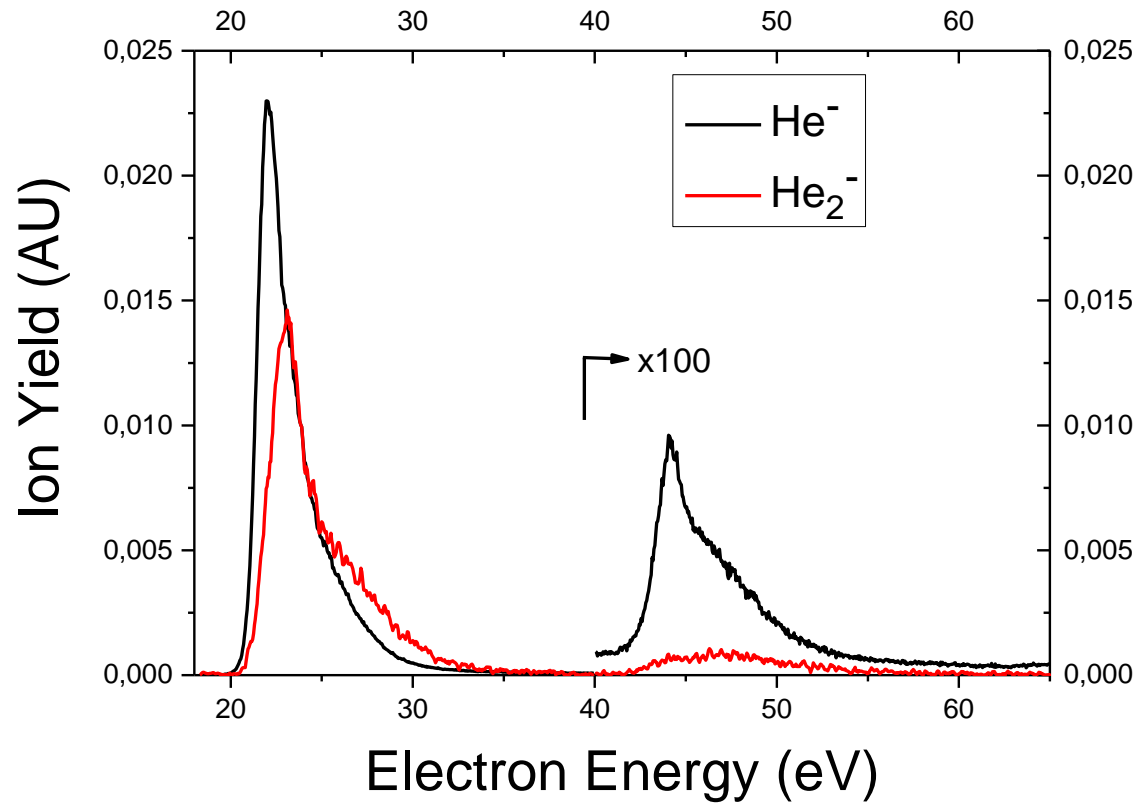
[3] J. Hiby, Ann. Phys (N.Y.) **34**, 473 (1939)

[4] Y. K. Bae et al., Phys. Rev. Lett. **52**, 747 (1984)

[5] T. Brage and C. Froese Fischer, Phys. Rev. A **44**, 72 (1991)

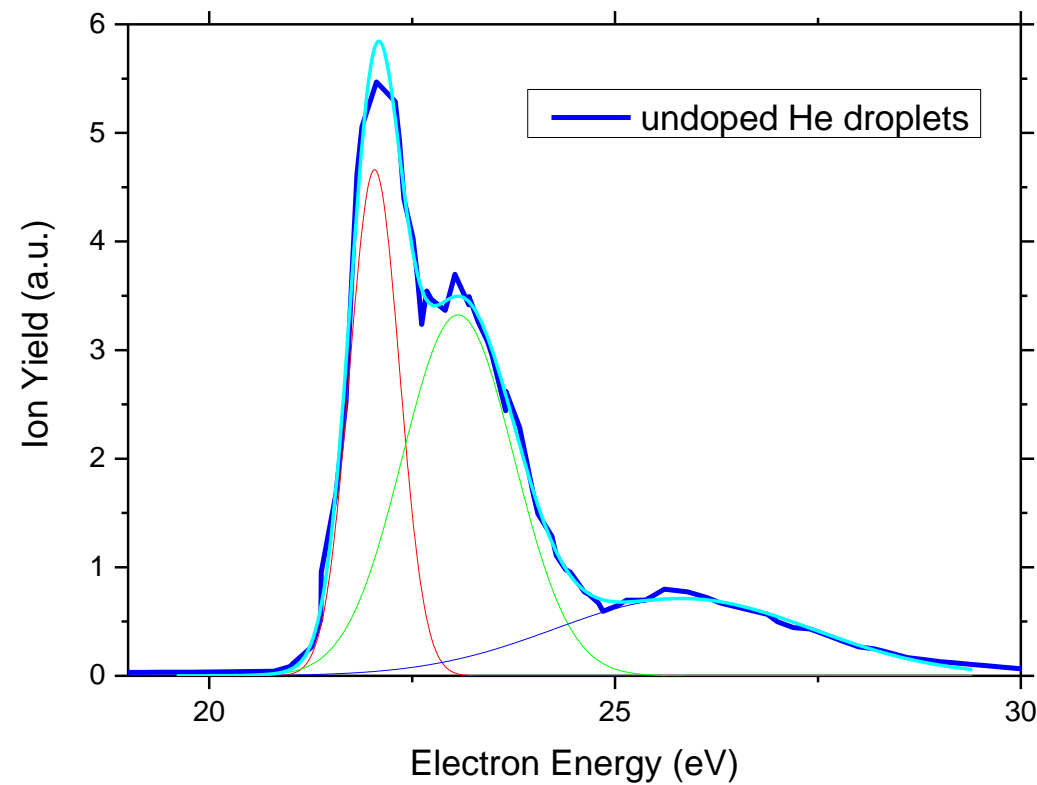
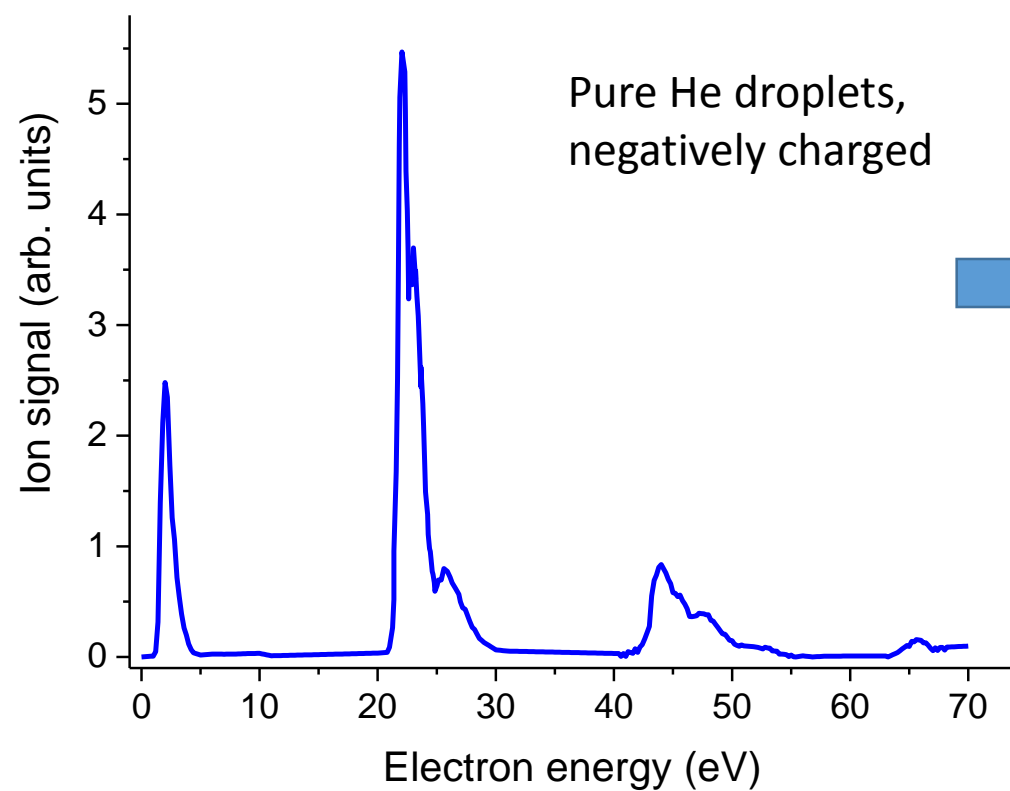
[6] S. E. Huber et al., Mol. Phys., **112**:5-6, 794-804 (2014)

Formation



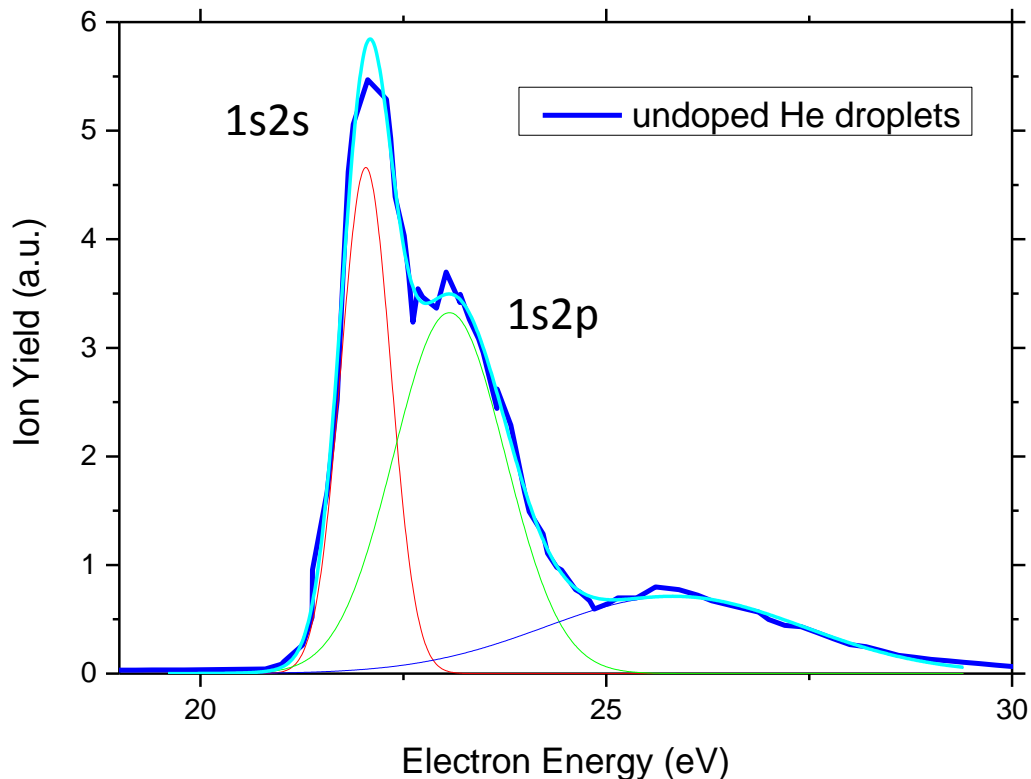
[1] U. Henne et al., J. Chem. Phys. **108**, 9327 (1998).

Formation

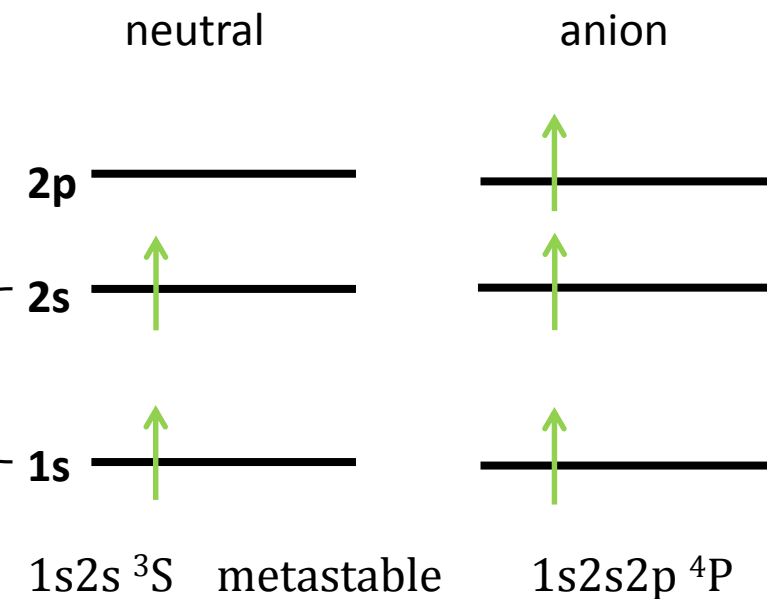


[1] U. Henne et al., J. Chem. Phys. **108**, 9327 (1998).

Formation



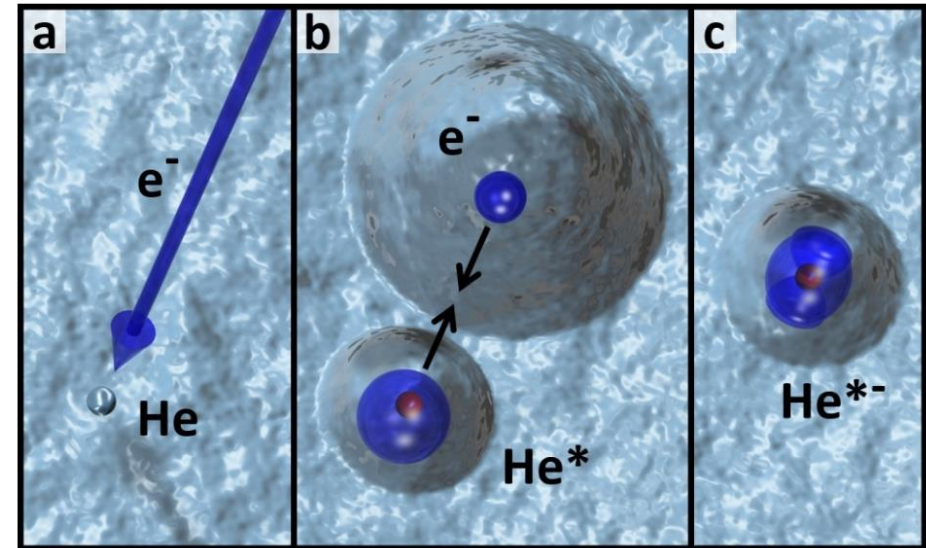
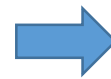
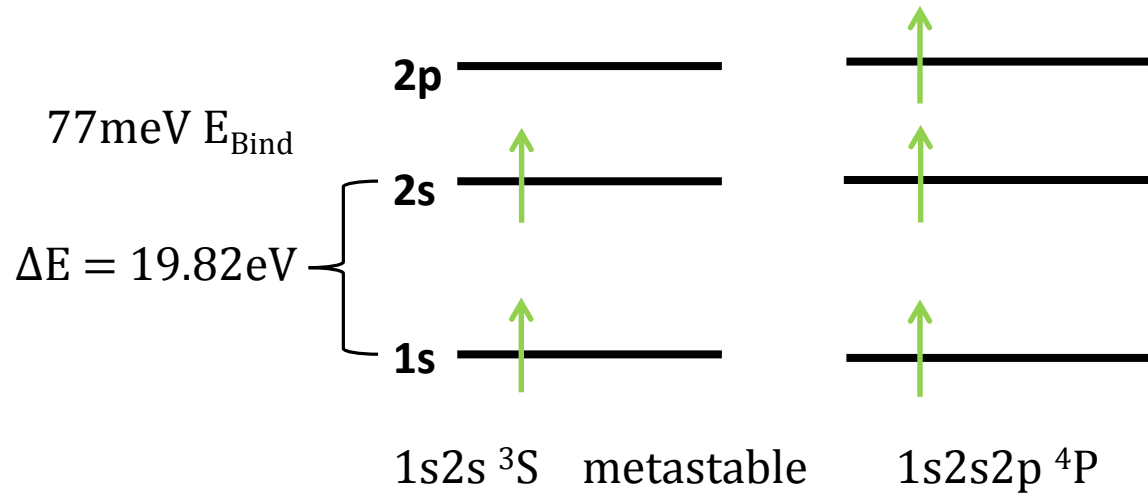
$\Delta E = 19.82 \text{ eV}$



- 19.7 eV \neq 22 eV 2.3 eV missing!
- Several other effects:
 - Helium barrier for the e⁻: 1.25 eV
 - Blueshift due to interaction with He: 0.80 eV

[1] U. Henne et al., J. Chem. Phys. **108**, 9327 (1998).

Formation

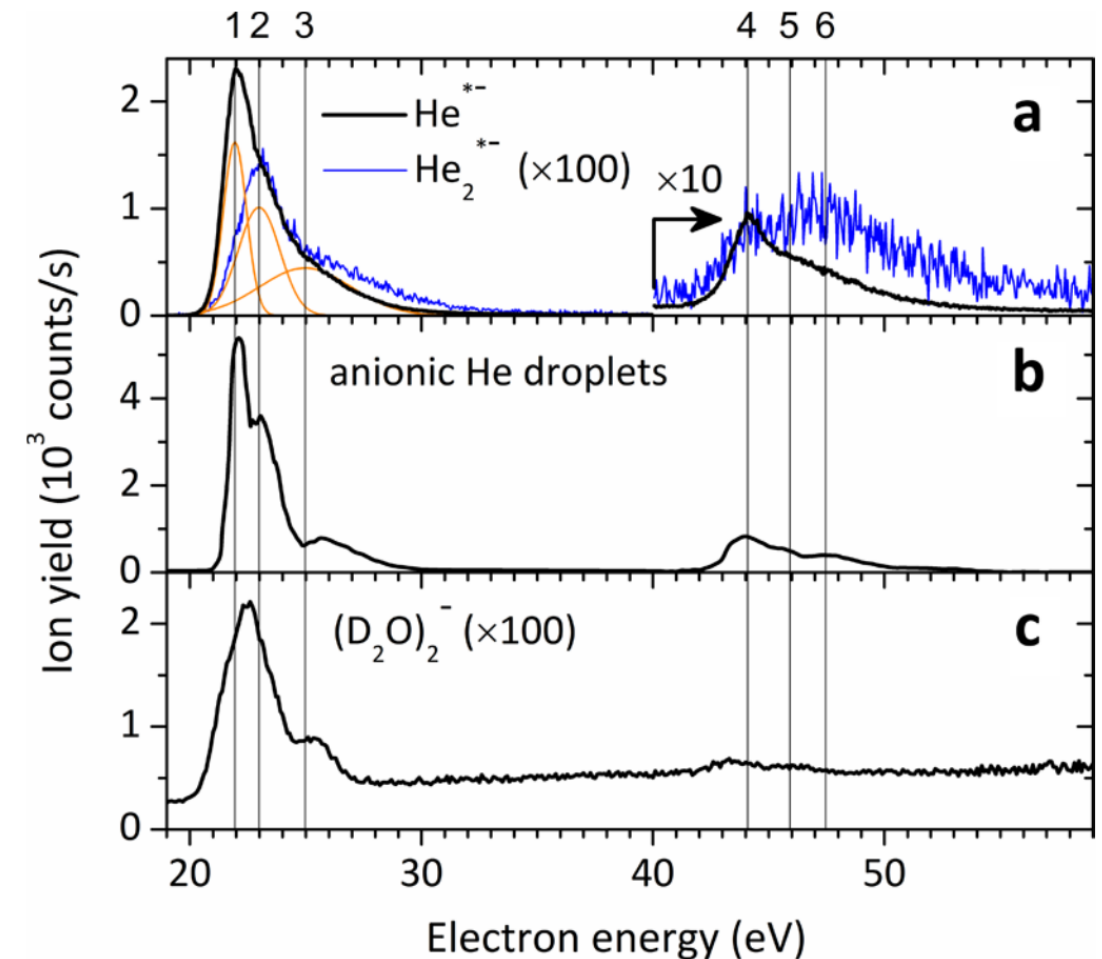


- 19.7 eV \neq 22 eV 2.3 eV missing!
- Several other effects:
 - Helium barrier for the e^- : 1.25 eV
 - Blueshift due to interaction with He: 0.80 eV¹

- e^- scatters with He, exciting it to one of three states
- slowed-down e^- forms bubble (as well as the excited He*)
- e^- attaches to He* and bubbles combine, because they gain energy in that process

Importance

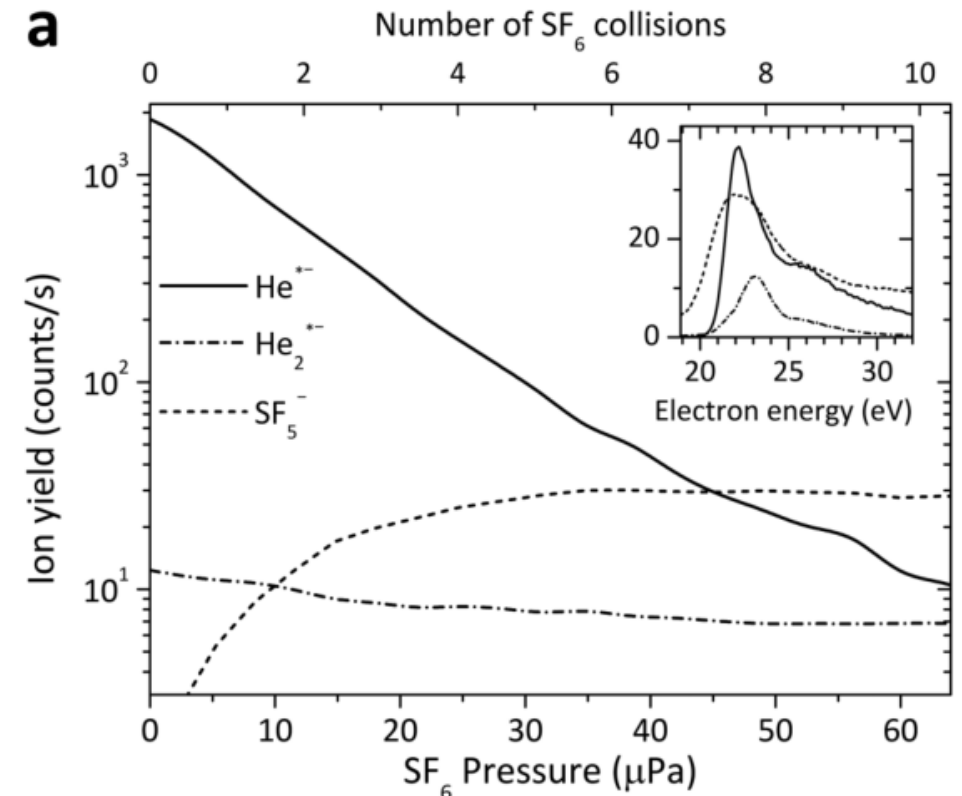
- New ionization channels in addition to the aforementioned ones:
 - $\text{He}^{*-} + X \rightarrow X^+ + \text{He} + e^-$
 - $\text{He}^{*-} + X \rightarrow X^{2-} + \text{He}^+$
 - $\text{He}^{*-} + \text{AB} \rightarrow \text{A}^- + \text{B} + \text{He}$
- He^{*-} might be the exotic fast negative charge carrier¹



Properties

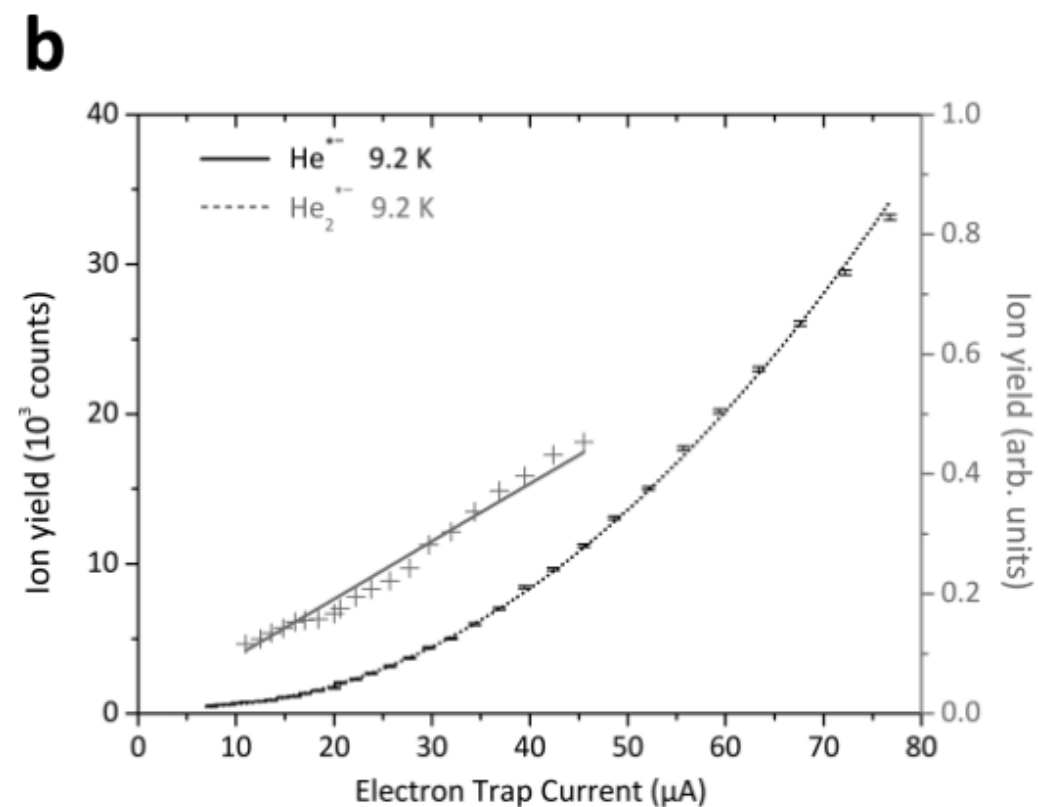
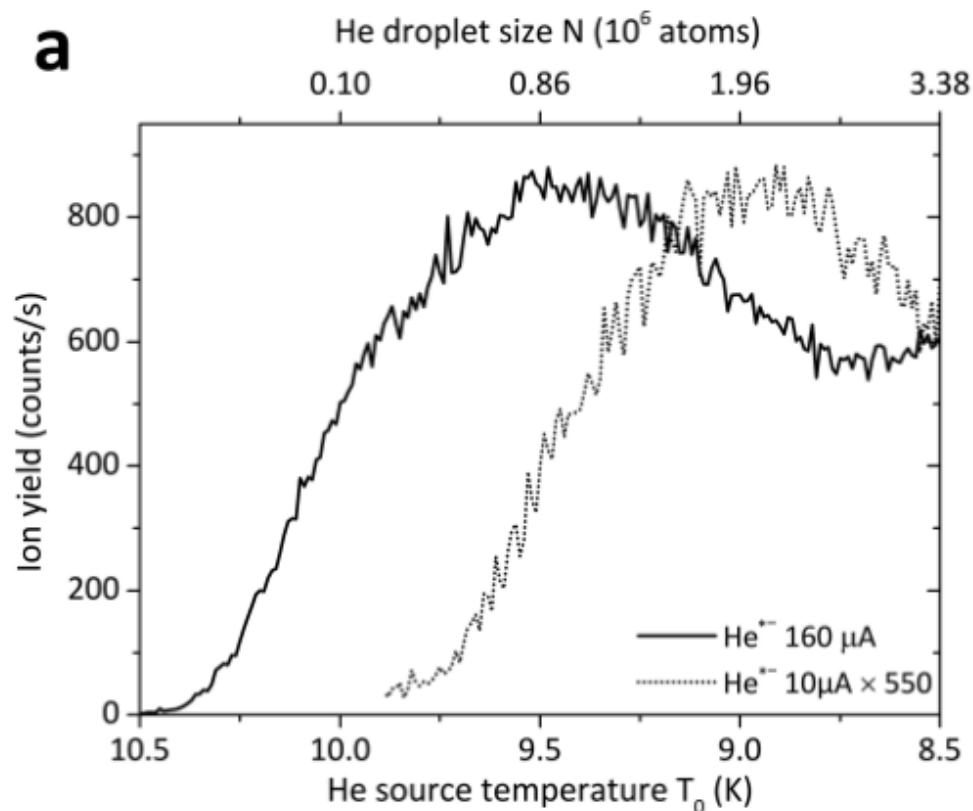
- CCSD(T) calculations ¹
- He^{*-} heliophilic, whereas He₂^{*-} is heliophobic
- Higher mobility of He^{*-}

	E_{Gain} (meV)	E_{Cost} (meV)	$E_{\text{Gain}} / E_{\text{Cost}}$
He	36.53	15.87	2.30
He [*]	70.74	78.07	0.91
He ^{*-}	211.47	76.97	2.75
He ₂ ^{*-}	14.49	46.50	0.31



Ejection?

- So if He^{*-} is heliophilic, why does it leave the droplet?



Summary

- It is possible to create He^{*-} with electron impact in HND
- He^{*-} is highly mobile and might be the exotic fast negative charge carrier in liquid helium
- He^{*-} might add new ionization mechanisms
- He_2^{*-} stays at the surface and is not involved in ionization processes

Thank you for your attention!



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